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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/549,950	09/20/2005	Charles H. Winter	WSU0200PUSA	1364
22045 7590 06/02/2009 BROOKS KUSHMAN P.C. 1000 TOWN CENTER TWENTY-SECOND FLOOR SOUTHFIELD, MI 48075				
EXAMINER				
ZIMMER, ANTHONY J				
ART UNIT		PAPER NUMBER		
1793				
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**Please find below and/or attached an Office communication concerning this application or proceeding.**

The time period for reply, if any, is set in the attached communication.

### Office Action Summary

**Application No.**

10/549,950

**Applicant(s)**

WINTER ET AL.

**Examiner**

ANTHONY J. ZIMMER

**Art Unit**

1793

**-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --**  
**Period for Reply**

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

**Status**

- 1) ☒ Responsive to communication(s) filed on 20 February 2009.
- 2a) ☒ This action is **FINAL**. 2b) ☐ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

**Disposition of Claims**

- 4) ☒ Claim(s) 1-3, 6-11, 13, 14, 25, 28-36 and 60-71 is/are pending in the application.
- 4a) Of the above claim(s) \_\_\_\_\_ is/are withdrawn from consideration.
- 5) ☐ Claim(s) \_\_\_\_\_ is/are allowed.
- 6) ☒ Claim(s) 1-3, 6-11, 13-14, 25, 28-36, and 60-71 is/are rejected.
- 7) ☐ Claim(s) \_\_\_\_\_ is/are objected to.
- 8) ☐ Claim(s) \_\_\_\_\_ are subject to restriction and/or election requirement.

**Application Papers**

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on \_\_\_\_\_ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.  
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).  
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) ☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

**Priority under 35 U.S.C. § 119**

- 12) ☐ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☐ All b) ☐ Some \* c) ☐ None of:
1. ☐ Certified copies of the priority documents have been received.
  2. ☐ Certified copies of the priority documents have been received in Application No. \_\_\_\_\_.
  3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

\* See the attached detailed Office action for a list of the certified copies not received.

**Attachment(s)**

- 1) ☒ Notice of References Cited (PTO-892)
- 2) ☐ Notice of Draftsperson's Patent Drawing Review (PTO-948)
- 3) ☐ Information Disclosure Statement(s) (PTO/SB/08)  
Paper No(s)/Mail Date \_\_\_\_\_
- 4) ☐ Interview Summary (PTO-413)  
Paper No(s)/Mail Date \_\_\_\_\_
- 5) ☐ Notice of Informal Patent Application
- 6) ☐ Other: \_\_\_\_\_

## **DETAILED ACTION**

### ***Claim Objections***

Claim 63 is objected to under 37 CFR 1.75(c), as being of improper dependent form for failing to further limit the subject matter of a previous claim. Applicant is required to cancel the claim(s), or amend the claim(s) to place the claim(s) in proper dependent form, or rewrite the claim(s) in independent form. Claim 63 lists heteroatom donor ligands that are not in the exclusive list presented in the claim from which claim 63 depends.

Claims 1, 25, and 60 are objected to because of the following informalities: The claims contain the redundant phrase "selected from the group consisting of selected from the group consisting of." Appropriate correction is required.

Claims 2, 3, 14, 36, 61, 64, and 71 objected to because of the following informalities: The claims should be formatted with proper Markush language, i.e. "selected from the group consisting of." Appropriate correction is required.

### ***Claim Rejections - 35 USC § 103***

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

The factual inquiries set forth in *Graham v. John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:

1. Determining the scope and contents of the prior art.
2. Ascertaining the differences between the prior art and the claims at issue.
3. Resolving the level of ordinary skill in the pertinent art.
4. Considering objective evidence present in the application indicating obviousness or nonobviousness.

**Claims 1-3, 6-10, 13-14, and 60-69 are rejected under 35 U.S.C. 103(a) as being unpatentable over the journal article by Wang et al. in view of Fan et al. "Preparation of Cu-Al<sub>2</sub>O<sub>3</sub> nano-composite powders by electroless copper plating."**

In regard to claims 1-3 and 60-64, Wang teaches ZnO, NiO, and SnO<sub>2</sub> nanoparticles (i.e. with an average number of oxygen atoms that are 1, 1, and 2 times the number of metal atoms respectively and having metal oxidation states of +2, +3, and +4 respectively). See Table 1 of Wang.

Wang is silent in regard to the total number of M atoms in each nanoparticle.

However, the nanoparticles of Wang have the same composition as the claimed nanoparticles and have a size falling in the range of the instant claims. Thus, the number of metal atoms would fall within the broad range required by the claims.

Wang does not teach the heteroatom donor ligands required by the claims. However, it would have been obvious to one of ordinary skill in the art to modify Wang with Fan because Fan teaches using 2,2'-bipyridyl (2,2'-bipyridine) to stabilize nano-sized powders. See abstract of Fan. After calcination (see Experimental section of

Wang), one of ordinary skill in the art would have been motivated to use the stabilizer (dispersing agent) of Fan in order to produce a stable dispersion of the nanoparticles of Wang in order to dispose the nanoparticles on a substrate or perform other uses requiring a stable dispersion. In general, selection of a particular dispersing agent among known dispersing agents (known to stabilize nanoparticle dispersion) is a matter of design choice and routine optimization in the absence of unexpected results.

In regard to claims 6-8 and 65-67, Wang teaches average particle sizes in the range of the claim(s). See table 1.

In regard to claims 9-10 and 68-69, Wang teaches spherical, ellipsoidal, and polyhedral morphologies and crystalline domains. See Figure 2 and associated text.

In regard to claims 13-14, when using the stabilizer, in addition to the bound molecules, loosely bound molecules in the solution thereof would also be present.

**Claims 25, 28-36, and 60-71 are rejected under 35 U.S.C. 103(a) as being unpatentable over the journal article by Aslam et al in view of Fan et al. "Preparation of Cu-Al<sub>2</sub>O<sub>3</sub> nano-composite powders by electroless copper plating."**

In regard to claims 25 and 60-64 Aslam teaches copper oxide (Cu<sub>2</sub>O) nanoparticles containing elemental copper in ratios of elemental copper to said copper oxide of 1:1, 1:3.5, and 1:3.5. See table 1. Using the notation of the instant claims, the composition of such particles can be represented on average as Cu<sub>3</sub>O, Cu<sub>16</sub>O<sub>7</sub>, and Cu<sub>16</sub>O<sub>7</sub>. Thus the number of O atoms is at least 0.01 times the number of copper atoms.

Aslam is silent in regard to the total number of M atoms in each nanoparticle.

However, the nanoparticles of Aslam have the same composition as the claimed nanoparticles and have a size falling in the range of the instant invention. (See the 102/103 rejection of claims 6-8 and 28-30 below). Thus, the number of copper atoms would fall within the broad range required by the claims. Also, the method of preparing the nanoparticles in Aslam is substantially similar to the instant method, and thus the processes would produce substantially similar products. In particular, both processes react metal ions with heteroatom donor ligands (Aslam reacts copper chloride and one of three capping agents, i.e. donor ligands) and then reduces the product thereof. See Synthesis section on page 80 of Aslam. See also MPEP 2112.01. Said nanoparticles of Aslam are capped (i.e. one or more heteroatom ligands are bonded to the surface of the nanoparticles). See Experimental section on page 80.

Aslam does not teach the heteroatom donor ligands instantly claimed.

However, it would have been obvious to one of ordinary skill in the art to modify Aslam with Fan because Fan teaches using 2,2'-bipyridyl (2,2'-bipyridine) to stabilize nano-sized copper-containing powders. See abstract of Fan. After production (see Experimental section of Aslam), one of ordinary skill in the art would have been motivated to use the stabilizer of Fan in order to produce a stable dispersion of the nanoparticles of Aslam in order to dispose the nanoparticles on a substrate or perform other uses requiring a stable dispersion. In general, selection of a particular dispersing agent among known dispersing agents (known to stabilize nanoparticle dispersion) is a matter of design choice and routine optimization in the absence of unexpected results.

In regard to claims 28-30 and 65-67, Aslam teaches particle sizes of 4-7 nm. See conclusion section on page 89, figures 5 and 6, and Table 1.

In regard to claim 31 and 68, Aslam teaches spherical, tubular (rod-shaped), and hexagonal (polyhedral) faceting shapes. See table 1 and Figures 5 and 6.

In regard to claim 32 and 69, Aslam teaches crystalline structure and amorphous structure. See Figure 6 and the text on pages 85-86.

In regard to claim 33-34 and 70-71, Aslam teaches a mixture of copper oxide ( $\text{Cu}_2\text{O}$ ) and elemental copper, i.e. comprising oxidation states 0 and +1. See Table 1.

In regard to claim 35-36, when using the stabilizer, in addition to the bound molecules, loosely bound molecules in the solution thereof would also be present.

**Claims 1-3, 6-11, 13-14, 25, 28-36, and 60-71 are rejected under 35 U.S.C. 103(a) as being unpatentable over the journal article by Dong et al in view of Fan et al.**

In regard to claims 1-3, 6-9, 11, 25, 28-31, 33-34, 60-68, and 70-71, Dong et al teaches copper oxide nanoparticles including:  $\text{Cu}_2\text{O}/\text{CuO}$  nanoparticles (thus the particles have a number of oxygen atoms that is at least 0.01 times the number of metal atoms), with a size of 34 nm, containing oxidation states of +1 and +2, and has cubic (polyhedral) morphology. See Table 1 of Dong.  $\text{Cu}_2\text{O}$  particles (thus the particles have a number of oxygen atoms that is at least 0.01 times the number of metal atoms and an oxidation state of +1) with sizes in the range of the claim(s) and morphologies of cubic (polyhedral) and spherical.

Dong is silent as to the total number of M (copper) atoms in each nanoparticle.

However, the nanoparticles of Dong have the same composition as the claimed nanoparticles (see above), and have a size falling in the range of the instant claims. Thus, the number of metal atoms would fall within the broad range instantly claimed.

Dong does not teach the heteroatom donor ligands instantly claimed.

However, it would have been obvious to one of ordinary skill in the art to modify Dong with Fan because Fan teaches using 2,2'-bipyridyl (2,2'-bipyridine) to stabilize nano-sized copper-containing powders. See abstract of Fan. After production (see Experimental section of Dong), one of ordinary skill in the art would have been motivated to use the stabilizer of Fan in order to produce a stable dispersion of the nanoparticles of Dong in order to dispose the nanoparticles on a substrate or perform other uses requiring a stable dispersion. In general, selection of a particular dispersing agent among known dispersing agents (known to stabilize nanoparticle dispersion) is a matter of design choice and routine optimization in the absence of unexpected results.

In regard to claims 10, 32, and 69, Dong teaches a product having crystalline and amorphous domains. See Figure 2 and associated text.

In regard to claims 13-14 and 35-36, when using the stabilizer, in addition to the bound molecules, loosely bound molecules in the solution would also be present.

### ***Response to Arguments***

Applicant's arguments with respect to the claims presented have been considered but are moot in view of the new ground(s) of rejection.



***Conclusion***

Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire **THREE MONTHS** from the mailing date of this action. In the event a first reply is filed within **TWO MONTHS** of the mailing date of this final action and the advisory action is not mailed until after the end of the **THREE-MONTH** shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than **SIX MONTHS** from the date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to ANTHONY J. ZIMMER whose telephone number is (571)270-3591. The examiner can normally be reached on Monday - Friday 7:30 AM - 5:00 PM EST.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Stanley Silverman can be reached on 571-272-1358. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Art Unit: 1793

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

ajz

/Steven Bos/  
Primary Examiner, Art Unit 1793